

## Water And Aqueous Systems Guided Answers Chemistry

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~~Systems". The Water Molecule: a Review. • Water is a simple tri-atomic molecule,~~

~~H. 2. O. •Each O-H bond is highly polar, because of the high electronegativity of the oxygen (N, O, F, and Cl have high values) •bond angle of water = 105o.~~

~~Chapter 15 Water and Aqueous Systems~~

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~~surface tension and vapor pressure of water. Water in the Liquid State (pages~~

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CHAPTER 15, Water and Aqueous Systems (continued) 6. Circle the letter next to each sentence that describes a result of the surface tension of water. a. In a full glass of water, the water surface seems to bulge over the rim of the glass.

## ~~SECTION 15.1 WATER AND ITS PROPERTIES (pages 445–449)~~

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