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2017 Page 3 of 13 Question 4

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Answer: C Explanation: 3KOH
 $(\text{aq}) + \text{AlCl}_3 (\text{aq}) \rightarrow 3\text{KCl} (\text{aq})$
 $+ \text{Al}(\text{OH})_3 (\text{s})$ $3\text{Zn}(\text{NO}_3)_2$
 $(\text{aq}) + 2\text{K}_3\text{PO}_4 (\text{aq}) \rightarrow \text{Zn}_3$
 $(\text{PO}_4)_2 (\text{s}) + 6\text{KNO}_3 (\text{aq})$
 $\text{Sr}(\text{OH})_2 + 2\text{NH}_4\text{CH}_3\text{COO}$
 $(\text{aq}) \rightarrow 2\text{NH}_4\text{OH} + \text{Sr}(\text{CH}_3\text{COO})$
 $_2 (\text{aq})$ $\text{Na}_2\text{SO}_4 (\text{aq}) +$

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$\text{Pb}(\text{NO}_3)_2 (\text{aq}) + \text{PbSO}_4 (\text{s}) + 2\text{NaNO}_3 (\text{aq})$ Question 5

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Explanation: Use kinematics equations to find the initial vertical velocity,

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given $v_{\text{vert}} = 0$ at 1.8 sec
after launch: $v = u + at$
 $0 = u - 10 \times 1.8$ $u = 18 \text{ms}^{-1}$. Then,
use trigonometry to determine
 v at 35° : $v = 18 \cos(55)$
 $= 31.4 \text{ms}^{-1}$. Question 18.
Answer: $R = 46 \text{ m}$ Explanation
: $v_{\text{horiz}} = v \times \cos(35) = 25.7 \text{ms}^{-1}$.

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$$x = v_h \times t = 46.26 \text{ m.}$$

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- Multiple Choice (1 mark each) Question 1 Answer: C
Explanation: The density of lines at a specific location in space reveals information about the strength of ...

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SOLUTIONS SECTION A - Core
Area of Study 1 - Motion ...
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Page 5 of 21 Question 12
Answer: $6.7 \times 10^4 \text{ N m}^{-1}$
Explanation: The

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gravitational potential energy at 5 m ($mgh = 0.15 \times 10 \times 5 = 7.5 \text{ J}$) is all converted to kinetic ...

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SOLUTIONS SECTION A - Core
Area of Study 1- Electric
Power Question 1 Answer: B

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2010 Page 3 of 16 Question
10 Answer: 300 turns

Explanation: Current at the
generator is $10000/250 = 40$

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A. Current in the lines is
10 A. Thus the transformer
is a step-up in the ratio of
1:4 ...

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of questions to be answered
Number of marks A 20 20 20 B
6 6 66 Total 86 Students are
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2017 Page 2 of 13 SECTION A:

Multiple-choice questions (1
mark each) Question 1

Answer: D Explanation: Ice
is less dense than water as

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the regular array of water molecules formed as a result of hydrogen bonding causes gaps to form. Question 2

Answer: C Explanation: 2140

$$J = 2.14 \text{ kJ n (water) =}$$

$$2.14/6.0 = 0.357 \text{ mol m}$$

$$\text{(water)} = 0.357 \times 18.0 =$$

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6.42 g Question 3 Answer: D

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SECTION A Area of Study
1-Nuclear Physics and
Radioactivity Question 1

Answer:A Explanation: As per
definition of each type of

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radioactive decay. Alpha is the heaviest and doubly

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