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5||STATES OF MATTER||PART 4||

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preparation of chlorine Chemistry

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T2 = final temperature (K).

Chemistry: Form WS6.4.1A. Name

... 4. 4.5 L. 1.035 atm. 375 K.

1.100 atm. 350. K. 5. 0.980 atm.

30.0°C. 185 mL. 0.900 atm.

28.0°C. 6 . WS6-4-1A_Using_the_C
ombined_Gas_Law.pdf

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Chemistry: Form N6.4.1A Using

the Combined Gas Law PHASES

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in the ta ble below by using the

combined gas law to find the

unknown variable in each row.

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Initial Final volume pressure
temperature volume pressure
temperature 1. 80.0 mL 96 kPa
27°C 99.0 kPa 0.0°C

Using the Combined Gas Law -
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Chem 1a Exam 1

Chemistry: Form WS6.3.1A Name _____

_____ PHASES OF MATTER Date _____

_____ Period _____ Phase Changes

Heating a substance in a given phase generally causes the temperature to increase. This causes the particles to move faster and collide harder. Harder, faster collisions cause particles to rebound harder moving them further apart. ...

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Chem 1a Exam 1

Revision Notes 4.1 The Periodic Table
4.2 Group 18 Elements 4.3 Group 1 Elements
4.4 Group 17

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Answers
Elements
4.5 The Periods and
Transition Metals Videos The
Group in Periodic Table
The Period in Periodic Table
Group 18 Elements
Group 1 Metal Reaction
of Alkali Metals with Oxygen
Gas
Group 17 Elements
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