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atoms, the s and p sublevels in the highest occupied
energy level are partially filled. Elements that are
good conductors of electric current and heat.

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row in the periodic table. question. group. answer.
vertical column in the periodic table. question.
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when elements are arranged in order of increasing
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according to atomic mass and used the arrangement to predict the properties of missing elements. elements that are characterize by the filling of p orbitals are classified as _____. which subatomic particle plays the greatest part in determining the properties of an element.

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To describe what oxidation and reduction mean to the chemist. To describe chemical reactions for which electrons are transferred (oxidation-reduction reactions).

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Label the diagram. Use these choices: covalent bond. hydrogen bond. slightly negative end. slightly positive end. 1. slightly negative end 2. slightly positive end 3. hydrogen bond 4. covalent bond In your textbook, read about mixtures with water.

Name

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accuracy and precision. Describe the five states of matter, and give an example of each

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