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Chapter 11 187 Exercise 11.3 –
Equation Stoichiometry: Iron is
combined with carbon in a series of

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reactions to form pig iron, which is about 4.3% carbon.

$$2\text{C} + \text{O}_2 \rightarrow 2\text{CO}$$
$$\text{Fe}_2\text{O}_3 + 3\text{CO} \rightarrow 2\text{Fe} + 3\text{CO}_2$$

2CO C (in iron) CO₂ Pig iron is easier to shape than pure iron, and the presence of carbon lowers its melting point

~~Chapter 11 - Gases~~

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Sample Study Sheet 11.2:

Abbreviated Electron Configurations

To get a review of the most important
topics in the chapter, fill in the blanks

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in the Key Ideas section. Work all of the selected problems at the end of the chapter, and check your answers with the solutions provided in this chapter of the study guide. Ask for help if you need it.

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11/4b FAQ. Q: What are hydrophobic and hydrophilic? A: Hydrophobic is a “fear of water”. Nonpolar substances are typically hydrophobic. They are not attracted to water. Hydrophilic on the other hand are “water lovers”. These compounds are attracted to water, and therefore

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