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Chemistry 3 (Practical) (Alternative A)  
12.00pm – 2.00pm (2nd Set) 1. A is 0.100  
mol dm<sup>-3</sup> solution of an acid. B is a solution  
KOH containing 2.8 g per 500 cm<sup>3</sup>. (a) Put  
A into the burette and titrate it against 25.0  
cm<sup>3</sup> portions B using methyl orange as  
indicator. Repeat the titration to obtain  
consistent titres.

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Chemistry 3 (Practical) (Alternative A)  
09.30am – 11.30am (1st Set) Chemistry 3  
(Practical) (Alternative A) 12.00pm –  
2.00pm (2nd Set) 1. A is 0.100 mol dm<sup>-3</sup>  
solution of an acid. B is a solution KOH  
containing 2.8 g per 500 cm<sup>3</sup>. (a) Put A into  
the burette and titrate it against 25.0 cm<sup>3</sup>  
portions B

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