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9.1: Molecular Shapes The Lewis electron-pair approach described previously can be used to predict the number and types of bonds between the atoms in a substance, and it indicates which

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atoms have lone pairs of electrons. This approach gives no information about the actual arrangement of atoms in space, however. 9.2: The VSEPR Model

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Molecular Shapes. • Lewis

structures give atomic

connectivity: they tell us which

atoms are physically connected to

which atoms. • The shape of a

molecule is determined by its

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• The angles made by the lines joining the nuclei of the atoms in a molecule are the bond angles.

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Molecular Shapes (Introduction)

The three-dimensional shapes and sizes of molecules are determined by their bond angles and bond lengths. Molecules with a central atom surrounded by atoms, denoted AX_n , adopt a number

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of different geometric shapes,
depending on the value of and on
the particular atoms involved.

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10 - Duration: 8:10. Mike
Christiansen 16,521 views. 8:10.
Molecular Orbital Theory I: The
Basic Idea - Duration: 5:07.

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In this video I'll teach you more about molecules' molecular geometries (shapes) and will further explain the difference between molecular geometry and elect...

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In this video I'll teach you how to use Lewis structures to predict a molecule's shapes and bond angles. I'll also teach you about VSEPR (Valence-Shell Elect...

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Lab 6 MOLECULAR GEOMETRY

Objectives Correlate Lewis dot structures with electron domain geometries View the effect of lone pairs of electrons on molecular. Molecular Polarity - Ms. Ose's Chemistry Website.

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Document. Gemensam strategi
mot nya hot - ett EU. Section 6-5.
Chemistry in Living Systems A
very basic overview Chapter 1.1
McGraw-Hill ...

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Major topics: VSEPR practice,
hybridization, sigma/pi bonds,
resonance cont'd, and para- vs.
diamagnetism.

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Chapter 9 Molecular Geometries

Access Free Chapter 9 Molecular Geometry And Bonding Theories. Chemistry, The Central Science, 11th edition Theodore L. Brown, H. Eugene LeMay, Jr., and Bruce E. Bursten John D. Bookstaver St. Charles Community College Cottleville, MO. Molecular Geometries and Bonding. © 2009, Prentice- Hall,

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Chapter 9 Molecular Geometries and Bonding Theories

Because the octahedral molecular geometry is symmetrical, the bond dipoles cancel, and the molecule is nonpolar, meaning

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that $\mu = 0$. Practice Exercise

Determine whether the following molecules are polar or nonpolar:

(a) NF_3 , (b) BCl_3 . Answers: (a) polar because polar bonds are arranged in a trigonal-pyramidal geometry, (b) nonpolar because

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Chemistry: The Central Science
Chapter 9: Molecular Geometry
and Bonding Theory The shape
and size of a molecule of a
particular substance, together
with the strength and polarity of

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Bonding, largely determine the properties of that substance o Change in shape could result in different properties 9.1: Molecular Shapes Lewis structures do not indicate the shape of the molecule The overall shape of a molecule is determined by its

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bond angles Central atom A us
bonded to n B atoms ...

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Valence Bond Theory • The

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Orbitals which overlap on adjacent atoms are either the standard atomic orbitals (s, p, d, f), or hybridized atomic orbitals made by combining individual atomic orbitals. • Each of the bonded atoms maintains its own atomic orbitals, but the electron

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- 9. Chapter 9: Molecular Geometry and Bonding Theories; 9.1: Molecular Shapes; 9.2: The VSEPR Model; 9.3: Molecular

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Shape and Molecular Polarity; 9.4:
Covalent Bonding and Orbital
Overlap; 9.5: Hybrid Orbitals; 9.6:
Multiple Bonds; 9.7: Molecular
Orbitals; 9.8: Second-Row
Diatomic Molecules; 9.E:
Exercises; 9.S: Molecular
Geometry and Bonding Theories

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Chapter 9: Molecular Geometry
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Recitation September 29th ALEKS
Objective 5 due September 27th

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ANGEL Quiz 5, September

29th Lecture 13: September

21st Lecture 14: September

23rd Lecture 15: September

26th Read: Ch. 2.6, 3.3-3.5,

9.1-9.3 Additional Preparation:

BLB 3: 25,29,35,37,51,53; Packet

3: 1-9 BLB 9: 23,25,27,38 ...

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PLAY. e domain. region about

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central atom where e are likely to be found, bonds or lone pairs. valence-shell e-pair (VSEPR) model. determines molecular geometries based on the repulsions b/n e domains domains orient themselves for LEAST repulsion.

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