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Compound \u0026  
its mixture †  
Lecture1 by Rahul  
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:ELEMENTS  
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CSIR TOPIC  
2-Separation of  
Mixture 3.2.1/3.2.2  
Describe the  
differences~~

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between elements,  
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PLAY. Boiling Point  
Elevation. the

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temperature  
difference between  
a solution's boiling  
point and a pure  
solvent's boiling  
point. Brownian  
Motion. the rapid  
movements of  
colloid particles that  
results from  
collisions of  
particles of the  
dispersion medium  
with the dispersed

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particles.

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solutions. STUDY.

PLAY. suspension.

mixture containing

particles that settle

if left un disturbed

ex: sand in water.



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colloid. a  
heterogeneous  
mixture of  
intermediate-sized  
particles ex: milk,  
nail polish.

Brownian motion.  
erratic movement of  
colloid particles.  
soluble.

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Solutions  
Mixtures and

Solutions. a type of

heterogeneous

mixture whose

particles settle out

over time and can

be separated from

the mixture by

filtration. a

heterogeneous

mixture of

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intermediate-sized  
particles (between  
atomic-size of  
solution particles  
and the size of

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Use the properties  
of seawater to  
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mixtures. Amy J.

Numerade Educator

03:59. Problem 2

Distinguish between

suspensions and

colloids. ...

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Section 14.1 Types  
of Mixtures pages

476 – 479 Section

Assessment 14.1

page 479 1. Explain

Use the properties

of seawater to

describe the

characteristics of



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mixtures. Answers will vary but might include that

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heterogeneous  
mixture whose  
particles settle out  
over time and can

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be separated from the mixture by filtration. a heterogeneous mixture of intermediate-sized particles (between atomic-size of solution particles and the size of suspension particles).

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Answer Effusion is  
the process that  
occurs when a gas  
escapes through a



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tiny hole in its container whereas diffusion is the tendency of molecules to move towards areas of lower concentration until the concentration is uniform throughout.

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