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2? A 2+ B 3+ C 2 – D 3 – 4. Fe. 2. O. 3 + x. CO . y. Fe + 3CO.
2. This equation will be balanced when. A . x = 1 and . y = 2 B . x = 2 and . y = 2 C . x = 3 and . y = 2 D . x = 2 and . y = 3. 5. An acidic solution contains A only hydrogen ions B only hydroxide ions C more hydrogen ions than hydroxide ions D more hydroxide ions than hydrogen ions. 6.

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The table below shows elements, atomic numbers and their melting points. The letters do not represent actual symbols of the elements.

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2 (g) $3\text{CO}_2(\text{g}) + 4\text{H}_2\text{O}(\text{g})$ Name the type of enthalpy change

which the student measured. No marks should be awarded for

'burning' since the word 'burned' appears in the text. (o) Unless

the question is clearly about a non-chemistry issue, eg costs in

industrial chemical process, a non-chemical answer gains no marks.

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2 by electricity / (electric) current / (flow of) electrons Mark independently b A = chlorine / Cl₂ Ignore Cl 3 B = hydrogen / H₂ Ignore H C = sodium hydroxide / NaOH Ignore references to sodium chloride If both name and formula given, both must be correct, but ignore Cl and H Award 1 mark for chlorine and hydrogen the wrong way round

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