

## 103 Section Assessment Chemistry Answers

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answer. 157.5 g 106.5 g 51 g Yes. Mass of reactants equals mass of products. Section 3.2 Assessment page 79 10. Classify each example as a physical change or a chemical change. a. crushing an aluminum can physical b. recycling used aluminum cans to make new aluminum cans physical c. aluminum combining with oxygen to form aluminum oxide chemical 11.

Matter—Properties and ChangesMatter—Properties and Changes

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Chapter Assessment

mass of oxygen 536.41 214.58 24.36 517.47 g 40% calcium, 12% carbon, 48% oxygen. 14.58/36.41 540.04% calcium 4.36/36.41 512.0% carbon 17.47/36.41 547.98% oxygen. Copyright © Glencoe/McGraw-Hill, a division of The McGraw-Hill Companies, Inc. 66Chemistry: Matter and Change • Chapter 4 Solutions Manual.

The Structure of the AtomThe Structure of the Atom

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